

L90 - Solid Evidence - LAB

anions:

cations:

Chemical reactions with ionic substances	Solid?
$\text{NaCl}(aq) + \text{KNO}_3(aq) \longrightarrow \text{NaNO}_3(aq) + \text{KCl}(aq)$	
$\text{NaCl}(aq) + \text{NaNO}_3(aq)$	
$\text{NaCl}(aq) + \text{KNO}_3(aq)$	
$\text{NaCl}(aq) + \text{NaNO}_3(aq)$	
$\text{Na}_2\text{CO}_3(aq) + \text{NaNO}_3(aq)$	
$\text{Na}_2\text{CO}_3(aq) + \text{KNO}_3(aq)$	
$\text{Na}_2\text{CO}_3(aq) + \text{NaNO}_3(aq)$	
$\text{Na}_2\text{CO}_3(aq) + \text{KNO}_3(aq)$	
$\text{NaOH}(aq) + \text{NaNO}_3(aq)$	
$\text{NaOH}(aq) + \text{Mg(NO}_3)_2(aq) \longrightarrow \text{NaNO}_3(aq) + \text{Mg(OH)}_2(s)$	
$\text{NaOH}(aq) + \text{Cu(NO}_3)_2(aq) \longrightarrow \text{NaNO}_3(aq) + \text{Cu(OH)}_2(s)$	
$\text{NaOH}(aq) + \text{AgNO}_3(aq) \longrightarrow \text{NaNO}_3(aq) + \text{AgOH}(s)$	

Please complete entire lab
Thank you!